



# Measurements Significant Figures Rounding

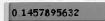




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There is no such thing as a perfect measurement

All measurements have errors Calculators display meaningless digits



Reporting  $10^{-10}$  values Data only good to  $10^{-1}$ 

Can you really measure 0.000000001 ml?



For this graduate Cylinder: Doubtful digit: tenth place (52.9 ml), so Any digit past tenths place is meaningless Any digit past tenths place is invalid Any digit past tenths place is fraud

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#### **Random Measurements**

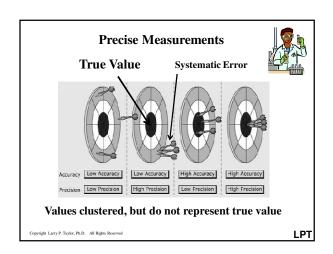


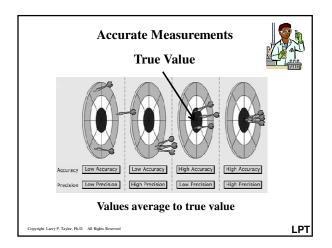
#### **True Value**

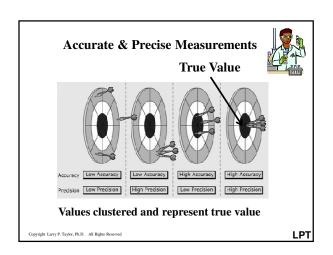
<b>\</b>				
Accuracy	Low Accuracy	Low Accuracy	High Accuracy	High Accuracy
	Low Procision	High Dendicion	Low Procision	High Procision

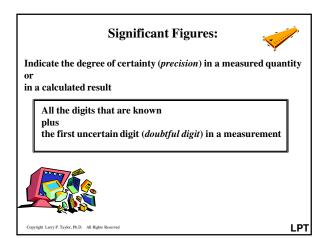
Values not clustered and do not represent true value

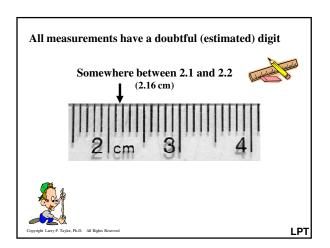
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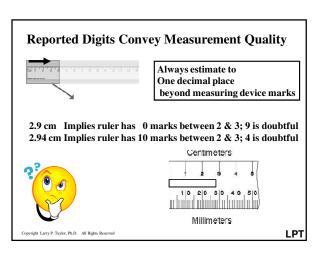












#### **Lab Measurements**



Distance (cm) measurements: 2 decimal digits



Mass measurements: all decimal digits



Volume (ml) measurements: 1 or 2 decimal digits (depends on cylinder markings)

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# **Never Round Data**



Always record digits displayed On instruments, last digit displayed is doubtful

If multiple measurements are used in calculations: errors in measurements are multiplied (propagated) magnitude of error may exceed ability to measure large error invalidates the experiment



Round Only after last calculation is completed

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# Certain & Doubtful Digits



	<u>Certain</u>	<u>Doubtful</u>
15.25 cm	15.2	5
894.22 g	894.2	2
1.7 L	1	7
36.94 mL	36.9	4
3.1 mg	3	1

Always last digit to the right in a measurement

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#### **Exact Numbers**

Numbers that are determined by counting or by definition ("per" expressions (conversion factors) are exact)

contain no uncertainty

have infinite  $(\infty)$  or unlimited significant figures.

Examples:

25 6 students 18 apples 491 cars 2.54 cm = 1 in 12 in = 1 ft 100 cm = 1 m 1000 g = 1 kg



Significant figures do not apply to exact numbers

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Quantity	Exact or Measured	Estimated Digit
6.27 meters		
15 students		
1.94 liters		
8,295 kg		
2.54 cm <b>=</b> 1 inch		
0.348 cm		
16		
45.0 mL		
1000 mg ≡ 1 g		

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Quantity	Exact or Measured	Estimated Digit
6.27 meters	Measured	7
15 students	Exact	
1.94 liters	Measured	4
8,295 kg	Measured	5
2.54 cm ≡ 1 inch	Exact	
0.348 cm	Measured	8
16	Exact	
45.0 mL	Measured	0
1000 mg ≡ 1 g	Exact	

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# **Significant Figures**

Begin with the first nonzero digit and end with the doubtful digit. The location of the decimal point has nothing to do with significant figures.

 $\begin{array}{ccc} Nonzero\ digits \\ 78.391\ km & (5\ figures) & 422.8\ gal\ (4\ figures) \end{array}$ 

 $\begin{array}{ccc} Leading\ zeros\ -\ never \\ 0.000391\ m\ \ (3\ figures) & 0.00255\ g\ \ (3\ figures) \end{array}$ 

Captive zeros 7,503 lbs

100,038 cm (6 figures) (4 figures)

Exact numbers -unlimited

200 cars 60 s **≡** 1 min 100 cm = 1 m

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# How many significant figures in:

18.043 m 1,000 km 1,000. mg 0.000667 L 90.800 tons



35 mL 0.0500700 g 9.360 x 10<sup>4</sup> s 8,628,000 cm 439.00 mL

 $1 \text{ mL} \equiv 1 \text{ cm}$ 



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# How many significant figures in:

18.043 m 1,000 km 1 1,000. mg 4 0.000667 L 3 90.800 tons 35  $35 \, mL$ 



0.0500700 g  $9.360 \times 10^4 \,\mathrm{s}$ 8,628,000 cm 439.00 mL 5  $1 \text{ mL} \equiv 1 \text{ cm}$ 



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#### Zeros

can be the doubtful digit can be used to show the decimal

use scientific notation to indicate significant figures:

ex: distance between the earth and the sun is 150,000,000 km

 $\begin{array}{lll} 1.5 \times 10^8 \ km & 2 \ significant \\ 1.50 \times 10^8 \ km & 3 \ significant \\ 1.500 \times 10^8 \ km & 4 \ significant \end{array}$ 



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# Indicate if the underlined zero is significant:

0.00807<u>0</u> km

<u>0</u> .050800 kg	<u>0</u> .008070 km
0. <u>0</u> 50800 kg	0. <u>0</u> 08070 km
0.05 <u>0</u> 800 kg	0.0 <u>0</u> 8070 km
0.0508 <u>0</u> 0 kg	0.008 <u>0</u> 70 km





 $0.05080\underline{0}$  kg

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#### Indicate if the underlined zero is significant:

<u>0</u> .050800 kg	n	<u>0</u> .008070 km  n
0. <u>0</u> 50800 kg	n	0. <u>0</u> 08070 km n
0.05 <u>0</u> 800 kg	y	0.0 <u>0</u> 8070 km n
0.0508 <u>0</u> 0 kg	y	0.008 <u>0</u> 70 km y
$0.05080\underline{0}~\mathrm{kg}$	y	0.00807 <u>0</u> km y





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# Re-write the quantity 40,000 mg

4 x 104 mg to show 1 sig. fig. to show 2 sig. fig.  $4.0 \times 10^4 \, mg$ to show 3 sig. fig. 4.00 x 10<sup>4</sup> mg 4.000 x 10<sup>4</sup> mg to show 4 sig. fig.



#### Re-write the quantity 9,340,000,000,000 ps

to show 3 sig. fig.  $9.34 \times 10^{12} \, \mathrm{ps}$ 9.340 x 10<sup>12</sup> ps to show 4 sig. fig. to show 5 sig. fig. 9.3400 x 1012 ps to show 6 sig. fig. 9.34000 x 10<sup>12</sup> ps

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#### **Rounding Off**



Calculators may contain digits that are not significant

If the first digit to be dropped is less than 5, leave the digit before it unchanged. If the first digit to be dropped is 5 or more, increase the digit before it by 1.

#### Round ONLY at the end of calculations

5.324657894

3 significant = 5.32

4 significant = 5.325 **5 significant = 5.3247** 

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#### **Problems**

Round 5.43 g 0.0448 m to 2 sig. figs.

5.4 g  $0.045~\mathrm{m}$ 

Round each of the following to 3 significant figures:

16.8477 L 16.8 L 5.6732 g 5.67 g 0.150 L 0.14986 L 861.85 kg 862 kg



 $4.203 \times 10^4 \text{ km}$  $4.20 \times 10^4 \, km$ 5.09810 x 10<sup>-3</sup> mm 5.10 x 10<sup>-3</sup> mm 0.00318756 m 0.00319 m 0.09025011 cm<sup>3</sup> 0.0903 cm<sup>3</sup>

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#### **Calculations - Multiplication & Division**

Round answer to the smallest number of sig. fig. in any factor  $% \left( 1\right) =\left( 1\right) \left( 1$ 

$$4.62 \ m \ x \ 3.1 \ m \ = 14.322 \ m^2 \ (14 \ m^2)$$

248.37 in 
$$\times \frac{2.54 \text{ cm}}{\text{in}} = 630.8598 \text{ cm} (630.86 \text{ cm})$$

 $\frac{6.230~g}{8.12~mL}~x~5~=~3.836207~g\,/\,mL~(3.84~g\,/\,mL)$ 





Failure to round to correct number of digits: Major factor in loss of exam points!

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#### **Calculations - Addition & Subtraction**

Round off the answer to the first column that has a doubtful digit





Failure to round to correct number of digits: Major factor in loss of exam points!

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#### Calculate to the correct number of significant figures

$$\frac{2.2745 \text{ g}}{3.40 \text{ cm x } 8.1 \text{ cm x } 7.090 \text{ cm}} = 0.0116487 \text{ g/cm}^3 (0.012 \text{ g/cm}^3)$$

95.34 cm<sup>3</sup> x 
$$\frac{21.3 \text{ g}}{\text{cm}^3}$$
 = 2030.74 g (2030 g)

 $58.953 \text{ g x} \frac{\text{mL}}{0.877 \text{ g}} = 67.2212 \text{ mL} \quad (67.2 \text{ mL})$ 

 $\frac{36.0059 \text{ g}}{12.2 \text{ cm}^3} = 2.70721 \text{ g/cm}^3 (2.71 \text{ g/cm}^3)$ 

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#### **Calculation with Many Conversions**



Physiologists suggest that people "burn" 15.0 kcals climbing a flight of stairs and 5.0 kcals descending a flight of stairs. If 3500 kcals represents a pound of fat, how many pounds of fat are lost during a 15 week semester using the stairs (instead of an elevator) for a CEM 101 class that meets on the third floor twice a week?

20.0 kcal x 3 stories x 2 days x 15 weeks x pounds fat = ? lbs fat story week semester 3500 kcal semester







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#### **Calculation with Many Conversions**



Physiologists suggest that people "burn" 15.0 kcals climbing a flight of stairs and 5.0 kcals descending a flight of stairs. If 3500 kcals represents a pound of fat, how many pounds of fat are lost during a 15 week semester using the stairs (instead of an elevator) for a CEM 101 class that meets on the third floor twice a week?

 $\frac{20.0 \text{ kcal}}{\text{story}} \times \underbrace{\frac{3 \text{ stories}}{\text{day}}}_{\text{day}} \times \underbrace{\frac{2 \text{ days}}{\text{week}}}_{\text{week}} \times \underbrace{\frac{15 \text{ weeks}}{\text{semester}}}_{\text{3500 kcal}} \times \underbrace{\frac{15 \text{ lbs fat}}{\text{semester}}}_{\text{semester}} \times \underbrace{\frac{15 \text{ weeks}}{\text{3500 kcal}}}_{\text{semester}} \times \underbrace{\frac{15 \text{ weeks}}{\text{3500 k$ 



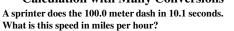




LP

# ¥

#### **Calculation with Many Conversions**





Start with Given and Wanted:

100.0 m 10.1 sec = ? mi / hr

"Cancel" units one at a time until left side unit = right side unit

 $\frac{100.0 \text{ m}}{10.1 \text{ sec}} \times \frac{100 \text{ cm}}{1 \text{ m}} \times \frac{1}{2.54} \times \frac{1 \text{ foot}}{12 \text{ in}} \times \frac{1}{5,280 \text{ ft}} \times \frac{60 \text{ sec}}{1 \text{ min}} \times \frac{60 \text{ min}}{1 \text{ hr}} = ? \text{ mi / hr}$ 

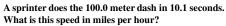
Units the same on both sides of equal sign; do the math:

 $\frac{100.0 \text{ m}}{10.1 \text{ sec}} \times \frac{100 \text{ cm}}{1 \text{ m}} \times \frac{1 - \text{inch x}}{2.54 \text{ cm}} \times \frac{1 \text{ foot}}{12 \text{ in}} \times \frac{1 - \text{mile}}{5,280 \text{ ft}} \times \frac{60 \text{ sec}}{1 \text{ min}} \times \frac{60 \text{ min}}{1 \text{ hr}} = 22.1479 \text{ mi/ hr}$ 

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<b>4</b>	
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#### **Calculation with Many Conversions**





Start with Given and Wanted:

100.0 m 10.1 sec

"Cancel" units one at a time until left side unit = right side unit

 $100.0 \text{ m} \times 100 \text{ cm} \times 1 \text{ inch} \times 1 \text{ foot} \times 1 \text{ mile} \times 60 \text{ sec} \times 60 \text{ min} = ? \text{ mi / hr}$ 10.1 sec 1 m 2.54 cm 12 in 5,280 ft 1 min 1 hr

Units the same on both sides of equal sign; do the math:

 $\frac{100.0 \text{ m}}{10.1 \text{ sec}} \times \frac{100 \text{ cm}}{1 \text{ m}} \times \frac{1}{2.54 \text{ cm}} \times \frac{1 \text{ foot}}{12 \text{ in}} \times \frac{1 \text{ mile}}{5,280 \text{ ft}} \times \frac{60 \text{ sec}}{1 \text{ min}} \times \frac{60 \text{ min}}{1 \text{ hr}} = 22.1479 \text{ mi/ hr}$ 

Round to 3 significant figures (dictated by the 10.1 sec term)

= 22.1 mi/ hr

= ? mi / hr

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#### **Calculation with Many Conversions**

Insurance statistics state that a person loses 8.00 minutes of average life for each cigarette smoked. Over the next 25 years, how much life is lost on average for a person smoking 0.5 pack (10 cigarettes) a day for 25 years? Express this loss in both minutes and years.

Start with Given and Wanted:

10 cigarettes

= ?min

"Cancel" units one at a time until left side unit = right side unit

 $\underline{10 \ cigarettes} \ x \ \underline{365.25 \ days} \ x \ \underline{8.00 \ min} \ x \ 25 \ years$ = ? min cigarette day year

Units the same on both sides of equal sign; do the math; round to 3 sig figs:

<u>10 cigarettes</u> x <u>365.25 days</u> x <u>8.00</u> min x 25 years = ? min year cigarette

Continue (without isolating) converting to years

 $x \frac{1}{60 \text{ min}}$   $x \frac{1}{24 \text{ hr}}$   $x \frac{1}{365.25 \text{ days}} = ? \text{ years}$ 



#### **Calculation with Many Conversions**

Insurance statistics state that a person loses 8.00 minutes of average life for each cigarette smoked. Over the next 25 years, how much life is lost on average for a person smoking 0.5 pack (10 cigarettes) a day for 25 years? Express this loss in both minutes and years.

Start with Given and Wanted:

10 cigarettes day

= ?min

"Cancel" units one at a time until left side unit = right side unit

 $\underline{10 \ cigarettes} \ x \ \underline{365.25 \ days} \ x \ \underline{8.00 \ min} \ x \ 25 \ years$ = ? min year cigarette day

Units the same on both sides of equal sign; do the math; round to 3 sig figs: <u>10 cigarettes</u> x <u>365.25 days</u> x <u>8.00</u> min x 25 years =  $7.31 \times 10^5$  min year cigarette day

Continue (without isolating) converting to years

 $7.31 \times 10^5 min \times \frac{1 \ hr}{60 \ min} \times \frac{1 \ day}{24 \ hr} \times \frac{1 \ day}{365.25 \ days}$ <u>year</u> = 1.39 years



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# **TI Specific Problem**



Some TI's auto delete trailing zeros 1.300 (correct number of sig figs) May display as 1.3 You must supply correct number of sig figs

For TI-84 Plus CE: Mode >> Answers Change "Auto" to "Dec" Hit Enter

BE CAREFUL
THIS MACHINE
HAS NO BRAIN

USE YOUR OWN

This will display all digits User selects correct sig figs

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#### All Calculations Must be Rounded to Appropriate Sig Figs





Failure to round to correct number of digits: Creates Undesirable Results!

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